INHIBITION OF RENAL TUBULAR TRANSPORT OF MORPHINE BY DIETHYLAMINOETHANOL IN THE CHICKEN*

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Abstract—Using the model system developed for renal tubular transport and metabolism, the morphine-morphine-3-ethereal sulfate (MES) system in the Sperber preparation of the chicken, diethylaminoethanol (DEAE) and possibly Lilly 18947 (2,4-dichloro-6phenylphenoxyethyldiethylamine) were shown to inhibit morphine transport. Because this inhibition with DEAE led to decreased access of morphine into the renal tubular cell, the excretion of morphine and MES in the urine decreased. However, the results of countercurrent distribution analyses indicated that the amount of MES relative to morphine did not decrease and, if anything, increased. Therefore, DEAE did not inhibit the metabolism of morphine to MES. Transport of ¹⁴C-tetraethylammonium was also inhibited; transport of 14C-MES was not altered by DEAE. Since Lilly 18947 reduced the apparent tubular excretion fraction of PAH and of administered ¹⁴C-MES, conclusions about the site of action of Lilly 18947 were not as firmly established. DEAE and possibly Lilly 18947 appeared to block morphine transport by virtue of their highly basic property. The fact that diethylaminoethanol occurs as a chemical moiety in these as well as in other compounds gives rise to the expectation that a number of compounds containing this moiety should show the same type of blocking activity on the organic cation transport system.

In a previous publication,¹ we reported that β -diethylaminoethyldiphenylpropylacetate (SKF 525A) and N-methyl-3-piperidyl-N',N'-diphenylcarbamate (MPDC) blocked the access of morphine into the renal tubular cell in the experiments in vivo with the Sperber preparation of the chicken. The attributes of this preparation, as developed for the study of the transport and metabolism of morphine,¹⁻⁵ made it possible to show clearly that SKF 525A and MPDC inhibited transport of morphine but did not inhibit the conjugation of morphine with sulfuric acid to form morphine-3-ethereal sulfate. Furthermore, this effect of SKF 525A and MPDC was shown to be due to their action as organic bases to block the organic cationic transport system. Therefore, this effect extended to blocking the transport of tetraethylammonium.

In the present study, Lilly 18947 (2,4-dichloro-6-phenylphenoxyethyldiethylamine) and diethylaminoethanol (DEAE) will be tested to see whether these compounds act like SKF 525A and MPDC on morphine transport. Lilly 18947 was included because, like SKF 525A, it is used as an inhibitor in drug metabolism studies. DEAE is a common chemical moiety which occurs not only in SKF 525A but also in Lilly 18947 as well as in CFT 1201 (diethylaminoethyl-2,2-diallyl-2-phenylacetate) and Sch 5712 (ethyl, diethylaminoethyl-2-butyl-2-ethylmalonate), which are also microsomal enzyme inhibitors. Because DEAE is the nitrogen-containing portion which confers the

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organic base property to all of these compounds, the present studies do give an insight into why many, if not all, such compounds will be inhibitors of the organic cationic transport system in the kidney.

METHODS

Renal tubular transport and metabolism of morphine were investigated in unanesthetized Rhode Island Red laying hens weighing 2·9-3·5 kg. They were prepared as described in the previous publication with SKF 525A. Plastic tubing with sponge rubber cuffs was sutured on the ureteral orifices for urine collection. The tubing was rinsed continuously at a rate of 0·25 ml/min with distilled water. Urine samples and rinse were collected for 10-min periods; the final volume of each sample was brought up to 10 ml by adding distilled water.

The control and test solutions for the intravenous infusion (rate of infusion = 0.42 ml/min) into the saphenous vein were made up in serial fashion. p-Aminohippuric acid (PAH), 16 mg, was dissolved in 200 ml of 0.9% sodium chloride. For the control infusion solution, N^{-14} CH₃-morphine hydrochloride (57 mc/m-mole), 0.3 to 1 ml stock solution of 10 μ c/ml, was added to 125 ml of the PAH solution. This control solution gave ¹⁴C counting ranges of 2623–6335 counts/min/0.1 ml. In each experiment, the infusion solution was counted to determine the exact ¹⁴C infusion rate. The ¹⁴C-morphine was obtained from Amersham/Searle. For the test infusion solution, either diethylaminoethanol (DEAE) or Lilly 18947 was dissolved in 42 ml of the control infusion solution. The doses of diethylaminoethanol and Lilly 18947 are given in the tables with the results.

In other experiments, either tetraethyl-l-[14 C]-ammonium (TEA) bromide, 3 mc/m-mole infused at 0·009 μ c/min (obtained from New England Nuclear Corp.), or N- 14 CH₃-morphine-3-ethereal sulfate (MES, 3·5 mc/m-mole infused at 0·003 μ c/min) was used in place of the 14 C-morphine and the effect of diethylaminoethanol and Lilly 18947 were assessed. The 14 C-MES was made biosynthetically by administering 14 C-morphine to the cat and isolating the crystalline metabolite as described previously. 8 - 8

The experimental protocol was to infuse the ¹⁴C control solution (morphine, MES or TEA) for 85–105 min before timed urine collections were started. This period allowed the system to come to a steady state. Then, 10-min control urine samples were collected simultaneously from both kidneys for five such periods. Subsequently, the intravenous infusion solution was switched to the test solution which included either the DEAE or Lilly 18947 in the control infusion solution. Ten-min urine samples were collected as before; seven or eight periods of collection were made.

Total ¹⁴C-radioactivity was determined by placing 0·1 ml of infusion solution or 0·2 ml of diluted urine solution in 15 ml of scintillation fluid prepared with 4 g 2,5-diphenyloxazole (PPO) and 50 mg 1,4-bis-2-(5-phenyloxazolyl)-benzene (POPOP) in a mixture of 1000 ml toluene and 500 ml Triton X-100. Details of the counting procedure were as described previously. PAH was determined colorimetrically by taking 0·5 ml of the diluted urine sample. ¹

The apparent tubular excretion fraction (ATEF) was calculated in the same fashion as previously where

$$\frac{EXC_{I} - EXC_{C}}{INF} \times 100 = ATEF$$

 $\rm EXC_I$ was the amount excreted in the urine from the infused side; $\rm EXC_C$ was the amount excreted from the contralateral side, and INF was the amount infused during each 10-min period. The infusion volume, dilution and sampling parameters chosen were such that ATEF for $^{14}\rm C$ could be calculated where the counts in 0·2 ml of diluted urine and 0·1 ml of $^{14}\rm C$ infusion solution could be inserted into the above equation to calculate the ATEF.

In all experiments where ¹⁴C-morphine was infused, the total radioactivity in the urine samples from the infused side was fractionated by countercurrent distribution. This fractionation enabled quantitation of ¹⁴C-morphine and its metabolite, morphine-3-ethereal sulfate. Two ml of diluted urine was adjusted to pH 8·5 with concentrated ammonium hydroxide with a microsyringe. The countercurrent distribution system consisted of 2 ml of each phase of a KHCO₃ (2 g/l) buffer, pH 8·5 and chloroform (with 1 %, v/v, N-butanol) in an 8-transfer manual operation as described previously. ¹ Chloroform was the mobile phase. The amount of MES was estimated by summing up the radioactivity in tubes 0–3. The amount of morphine was calculated as the total radioactivity in the countercurrent system minus that of the MES.

Significance of the differences between mean values was assessed by the *t*-test and, even though P values less than 0.01 were obtained in many cases, $P \le 0.05$ was taken as the level of significance.

RESULTS

Effects of Lilly 18947 on morphine and PAH transport. Table 1 shows the data for the ATEF of 14 C-morphine and PAH where high 14 C and PAH ATEF values during control periods indicate active transport. Profound decreases in 14 C ATEF (all P values ≤ 0.05) occurred when Lilly 18947 was infused at 0.2 to 0.5 mg/kg/min. The ATEF for PAH also fell significantly ($P \leq 0.05$) in two of the three experiments. However, this fall was not as great as the fall in the 14 C ATEF. The fall in PAH ATEF seen with SKF 525A and MPDC previously was ascribed to opening of the venous bypass valve in the renal portal system. By analogy, the effect of Lilly 18947 on the PAH ATEF may be on the venous bypass valve also. The quantitatively much greater drop in 14 C than in PAH ATEF must be due to an additional effect. The presence of this additional effect is readily seen by examining the 14 C/PAH ATEF ratio in Table 1. In all cases, Lilly 18947 caused a differentially greater fall in 14 C than in PAH ATEF, as seen by the fall in the ratio ($P \leq 0.05$).

The last column in Table 1 gives the results of the countercurrent analysis of the urine samples from the infused side. Note that the results are given in terms of the per cent of 14 C-radioactivity in the urine sample appearing as 14 C-MES. Since the total 14 C excretion was decreased by Lilly 18947 as given by the 14 C ATEF, the purpose of fractionating the excreted 14 C into MES and morphine was to see whether this decrease in 14 C excretion might be due to a decrease in the excretion of 14 C-morphine or 14 C-MES or both. If one expects Lilly 18947 to inhibit only the metabolism of morphine, then the per cent 14 C-MES must decrease. The data show that the per cent 14 C-MES does not decrease; the per cent 14 C-MES increased ($P \le 0.05$). Thus, the results are not compatible with any postulate involving inhibition of metabolism of morphine to MES. The hypothesis compatible with the results is that Lilly 18947, like SKF 525A and MPDC, may be inhibiting transport of 14 C-morphine into the renal tubular cell. In other words, the fall in 14 C ATEF is due not only to opening of the

Table 1. Effect of Lilly 18947 on ¹⁴C-morphine and PAH transport

		Dose		Apparen	Apparent tubular excretion fraction*	ction*	% of ¹⁴ C
Expt. No.	Treatment drug	(mg/kg/min)	No. 10-min periods	14C	РАН	14C/PAH	as ^{ĭ4} C-MES†
P48	Control 18947	0.5	5	40.0 ± 2.4 5.5 ± 2.3‡	$\begin{array}{c} 52.6 \pm 2.4 \\ 22.9 \pm 5.1 \\ \end{array}$	$\begin{array}{c} 0.76 \pm 0.03 \\ 0.25 \pm 0.05 \end{array}$	$65.2 \pm 0.9 \\ 82.6 \pm 2.3 \\ \pm$
P52	Control 18947	0.3	. v.∞	51.7 ± 1.4 13.9 ± 1.3 ‡	85.0 ± 2.5 53.6 ± 7.0	0.60 ± 0.03 $0.23 \pm 0.02 \pm$	53.6 ± 2.7 $76.4 \pm 2.9 \ddagger$
P62	Control 18947	0.2	% &	$\begin{array}{c} 60.0 \pm 0.6 \\ 21.2 \pm 2.4 \end{array}$	90.4 ± 4.7 80.8 ± 2.5	$\begin{array}{c} 0.67 \pm 0.03 \\ 0.27 \pm 0.03 \end{array}$	$62.5 \pm 1.6 \\ 79.0 \pm 1.4 \ddagger$

† Values given are means ±S.E. Countercurrent analysis was performed on each urine sample from the infused side and MES was measured as the ¹⁴C procurring in tubes 0-3; this ¹⁴C expressed as a percentage of the total ¹⁴C present in the system gave per cent ¹⁴C-MES. The per cent ¹⁴C-morphine would be * Values given are means ±S.E. Apparent tubular excretion fraction (ATEF) was calculated as given in Methods.

100—per cent ¹⁴C-MES. \ddagger Mean value for control vs drug treatment was significantly different (P \leq 0·05) by the Student's *t*-test.

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TABLE 2. EFFECT OF DIETHYLAMINOETHANOL (DEAE) ON 14C-MORPHINE AND PAH TRANSPORT

		Dose		Apparen	Apparent tubular excretion fraction*	action*	% of ¹⁴ C
Expt. No.	Treatment drug	(mg/kg/min)	No. 10-min periods	14C	РАН	14C/PAH	as ¹⁴ C-MES†
P28	Control DEAE	0.5	v, ∞	28·1 ± 2·1 15·5 ± 0·6‡	57.5 ± 2.3 63.4 ± 3.2	$\begin{array}{c} 0.49 \pm 0.03 \\ 0.25 \pm 0.02 \end{array}$	56·1 ± 2·7 75·8 ± 1·1‡
P32	Control DEAE	0.5	V ∩ ∞	$22.9 \pm 3.3 \\ 11.6 \pm 1.1 \ddagger$	$51.2 \pm 9.2 \\ 63.2 \pm 2.1$	$\begin{array}{c} 0.49 \pm 0.07 \\ 0.18 \pm 0.02 \end{array}$	$65.3 \pm 1.2 \\ 76.5 \pm 1.5 \ddagger$
P72	Control DEAE	0.5	√ 0 ∞	$\begin{array}{c} 33.0 \pm 2.5 \\ 5.2 \pm 1.6 \end{array}$	43.4 ± 3.4 31.2 ± 4.6	$\begin{array}{c} 0.76 \pm 0.02 \\ 0.23 \pm 0.09 \end{array}$	48.9 ± 1.2 67.1 ± 2.4
P75	Control DEAE	6.5	V ∩ ∞	33.8 ± 6.1 16.0 ± 1.3	44.4 ± 8.9 42.0 ± 1.8	$\begin{array}{c} 0.80 \pm 0.05 \\ 0.38 \pm 0.03 \end{array}$	$68.0 \pm 1.2 \\ 75.3 \pm 2.5 \ddagger$

* Values given are means ±S.E. Apparent tubular excretion fraction (ATEF) was calculated as given in Methods.
† Values given are means ±S.E. Countercurrent analysis was performed on each urine sample from the infused side and MES was measured as the ¹⁴C occurring in tubes 0-3; this ¹⁴C expressed as a percentage of the total ¹⁴C present in the system gave per cent ¹⁴C-MES. The per cent ¹⁴C-morphine would be 100 - per cent 14C-MES.

‡ Mean value for control vs drug treatment was significantly different (P ≤ 0.05) by the Student's t-test.

venous bypass valve but is due additionally to block of ¹⁴C morphine transport by Lilly 18947. These points will be pursued further in the Discussion.

Effects of diethylaminoethanol on morphine and PAH transport. Table 2 shows the results with DEAE which in a dose of 0.5 mg/kg/min reduced the 14 C ATEF during 14 C-morphine infusion. In contrast to Lilly 18947, DEAE had no effect on the ATEF of PAH (P > 0.05). Thus, the complicating effect of opening of the venous valve has been circumvented and the drop in 14 C ATEF indicates an effect on the morphine system alone. Therefore in all four cases there is a consistent fall in the 14 C-PAH/ATEF ratio (P \leq 0.05) produced by DEAE. The countercurrent analysis results in Table 2, last column, show that DEAE does not inhibit the metabolism of morphine because the per cent 14 C-MES did not drop. In fact, the per cent MES increased in all cases (P \leq 0.05). These results taken together with the fact that 14 C ATEF was reduced by DEAE were interpreted to indicate that DEAE inhibited transport but not metabolism of 14 C-morphine.

Effect of Lilly 18947 and diethylaminoethanol on ¹⁴C-TEA transport. Since the evidence up to this point indicated that both Lilly 18947 and DEAE appeared to be inhibiting transport of morphine, it was highly likely that such an inhibitory effect of Lilly 18947 and DEAE would extend to transport of bases other than morphine. TEA is known to be transported by the same cation system that transports morphine; ^{1,4} therefore, the effect of these compounds on TEA transport was studied.

Table 3 shows that Lilly 18947 produced significant reductions in the 14 C-TEA to PAH ATEF (P ≤ 0.05) where in both experiments the ratios were drastically reduced. Therefore, Lilly 18947 seems to block the transport of 14 C-TEA.

Expt.	Treatment	Dose	No. 10-min	Apparen	t tubular excret	tion fraction
No.	drug	(mg/kg/min)	periods	¹⁴ C-TEA	PAH	14C-TEA/PAH
P84	Control		5	26·1 ± 2·6	11·3 ± 2·0	2·69 ± 0·55
	18947	0.3	8	4·5 ± 4·2*	$18.9 \pm 2.3*$	$0.28 \pm 0.26*$
P104	Control		5	38.8 ± 2.6	$32\cdot3 \pm 2\cdot4$	1.20 ± 0.15
	18947	0.3	7	$1.5 \pm 2.8*$	$32\cdot6\pm7\cdot1$	$0.07 \pm 0.07*$

Table 3. Effect of Lilly 18947 on 14 C-tetraethylammonium (TEA) and PAH transport

Table 4 shows similar experiments in which DEAE was used in place of Lilly 18947. In these experiments, DEAE unequivocally reduced the ¹⁴C-TEA ATEF ($P \le 0.05$) in the absence of any reduction in PAH ATEF (in experiment P96, the PAH ATEF was significantly raised above the control value (P < 0.05). The significant reduction in all cases of the ¹⁴C-TEA/PAH ratio ($P \le 0.05$) indicated that DEAE blocked transport of ¹⁴C-TEA.

Effect if Lilly 18947 and diethylaminoethanol on ¹⁴C-MES transport. The possible effect of Lilly 18947 and DEAE on the transport of ¹⁴C-MES was tested. Table 5 shows that in three of the four experiments, Lilly 18947 reduced the ¹⁴C-MES ATEF ($P \le 0.05$). In these same three experiments, there was a tendency for the PAH ATEF

^{*} Mean value for control vs Lilly 18947 treatment was significantly different (P \leq 0.05) by the Student's t-test.

Expt.	Treatment	Dose	No. 10-min	Apparent tubular excretion fraction		
No.	drug	(mg/kg/min)	periods	¹⁴ C-TEA	PAH	14C-TEA/PAH
P94	Control DEAE	0.5	5 8	59·3 ± 1·9 27·1 ± 1·3*	48.8 ± 3.7 46.9 ± 1.3	1·13 ± 0·03 0·58 ± 0·02*
P96	Control DEAE	0.5	5 8	43·7 ± 5·7 4·4 ± 0·7*	$28.6 \pm 5.2 \\ 44.1 \pm 1.5*$	$\begin{array}{c} 1.60 \pm 0.10 \\ 0.10 \pm 0.02 \end{array}$
P100	Control DEAE	0.5	5 8	$35.0 \pm 2.6 \\ 13.7 \pm 1.4*$	28.4 ± 1.9 27.7 ± 1.7	1·24 ± 0·06 0·50 ± 0·06*

TABLE 4. EFFECT OF DIETHYLAMINOETHANOL (DEAE) ON 14C-TEA AND PAH TRANSPORT

to fall also. The ratio of the ¹⁴C-MES/PAH ATEF in all three cases tended to be depressed. Considering all four experiments, it is difficult to state whether Lilly 18947 had any consistent effect on ¹⁴C-MES transport.

Expt.	Treatment	Dose	No. 10-min	Apparent	tubular excretio	on fraction
No.	drug	(mg/kg/min)	periods	¹⁴ C-MES	PAH	14C-MES/PAH
P112	Control 18974	0.3	5 8	41·8 ± 0·05 22·9 ± 0·19*	53·1 ± 3·4 40·5 ± 5·0	0·79 ± 0·04 0·53 + 0·11*
P114	Control 18947	0.5	5 8	60.1 ± 0.1 $26.3 \pm 0.2*$	66·8 ± 0·1 33·4 ± 0·2*	0.90 ± 0.02 0.75 ± 0.08
P116	Control 18947	0.3	5 8	43.1 ± 2.8 $31.1 \pm 1.7*$	$47.7 \pm 2.5 \\ 36.7 \pm 5.7$	0·91 ± 0·06 0·75 ± 0·04*
P118	Control 18947	0.2	5 8	71.0 ± 2.4 72.7 ± 1.2	86·2 ± 7·5 92·3 ± 2·4	$0.85 \pm 0.07 \\ 0.79 \pm 0.03$

TABLE 5. EFFECT OF LILLY 18947 ON 14C-MES AND PAH TRANSPORT

In contrast to these results, the experiments in Table 6 clearly show that DEAE did not have any effect on ¹⁴C-MES transport. In all cases, no significant change in the ratio of the ¹⁴C-MES/PAH ATEF was found. In the one case (experiment P110) where the ¹⁴C-MES ATEF was reduced by DEAE, there was a corresponding decrease in PAH ATEF so that the ratio of the ATEF values was unchanged in going from control to DEAE treatment. Therefore DEAE had no effect on ¹⁴C-MES transport.

DISCUSSION

A brief review of the morphine-MES renal tubular transport and metabolism model system is necessary in order to interpret fully the present experiments. Morphine is transported by the renal tubular system that actively secretes organic cations.^{4,5} It is metabolized within the renal tubular cell to morphine-3-ethereal sulfate;^{4,5,8} this metabolism can be inhibited by catechol, thereby depressing the appearance of MES

^{*} Mean value for control vs DEAE treatment was significantly different ($P \le 0.05$) by the Student's *t*-test,

^{*} Mean value for control vs Lilly 18947 treatment was significantly different (P \leq 0.05) by the Student's t-test.

Expt.	Treatment	Dose	No. 10-min	Apparent	tubular excretion	on fraction
No.	drug	(mg/kg/min)	periods	14C-MES	PAH	¹⁴ C-MES/PAH
P106	Control DEAE	0.5	5 8	$\begin{array}{c} 29.0 \pm 15.6 \\ 19.9 \pm 6.0 \end{array}$	25·4 ± 10·15 15·7 ± 5·9	1.08 ± 0.08 1.16 ± 0.11
P108	Control DEAE	0.5	5 8	$\begin{array}{ccc} 42.4 \pm & 3.1 \\ 39.6 \pm & 2.0 \end{array}$	$38\cdot 1 \pm 3\cdot 1 41\cdot 1 \pm 1\cdot 4$	$\begin{array}{c} 1.12 \pm 0.06 \\ 0.97 \pm 0.04 \end{array}$
P110	Control DEAE	0.5	5 7	$64.9 \pm 2.3 55.2 \pm 2.7*$	57·9 ± 1·5 49·5 ± 1·9*	$\begin{array}{c} \textbf{1.12} \pm \textbf{0.01} \\ \textbf{1.12} \pm \textbf{0.06} \end{array}$

TABLE 6. EFFECT OF DIETHYLAMINOETHANOL (DEAE) ON 14C-MES AND PAH TRANSPORT

in the ipsilateral urine samples. At the same time, an expected compensatory increase in free ¹⁴C-morphine excretion occurs while the total ¹⁴C ATEF remains unchanged.² MES, which is administered by i.v. infusion, is transported by the system that secretes organic anions.⁴ Probenecid, a compound which blocks the anion transport system, blocks the transport of this administered MES but does not block excretion of MES, which is formed intracellularly by metabolism of morphine to MES. Therefore, a site of action of probenecid is placed at the peritubular border of the renal tubular cell.⁴ Many of these general concepts have been further substantiated by experiments with the renal tubular transport and metabolism studies with the serotonin–5-hydroxy-indoleacetic acid model system.³ Reviews of these studies are available.^{9,10} In a further paper, a site of action of SKF 525A and N-methyl-3-piperidyl-N',N'-diphenylcarbamate (MPDC) was shown to be on the transport of morphine and not on the metabolism of morphine in the kidney.¹

The present experiments indicate that DEAE blocks the organic cation transport system of the renal tubular cell (as did SKF 525A and MPDC¹). The evidence for this statement is derived from several different experiments. First, DEAE reduces the ¹⁴C ATEF during ¹⁴C-morphine infusion in the absence of any effect on PAH transport. Second, this reduction in ¹⁴C ATEF results from a fall in the concentration of both ¹⁴C-morphine and ¹⁴C-MES in the urine. Since the relative proportion of ¹⁴C-MES to ¹⁴C-morphine is larger during DEAE infusion than during control periods without DEAE, there exists no evidence that DEAE inhibits metabolism of morphine to MES. Third, DEAE does not inhibit the transport of ¹⁴C-MES that was administered; therefore, it is highly unlikely that intracellularly formed MES excretion would have been blocked. Fourth, DEAE blocks the transport of ¹⁴C-TEA without affecting PAH transport. Thus, we conclude that DEAE blocks morphine transport by virtue of its ability to block the organic cation transport system. One probable location of the cation transport system appears to be on the peritubular side.³

Certain other possibilities were eliminated by making the following considerations. In the face of a decreased delivery of ¹⁴C-morphine from the peritubular fluid to the intracellular compartment, one might question whether in the present experiments we would have been able to detect inhibition of metabolism if it were present simultaneously with the reduction in morphine delivery. Inhibition of metabolism of morphine to MES within the renal tubular cells is known to lead to a compensatory

^{*} Mean value for control vs DEAE treatment was significantly different ($P \le 0.05$) by the Student's t-test.

increase in the proportion of morphine relative to MES, as cited above on the work with catechol. The present results clearly show that the proportion of ¹⁴C-MES increased and ¹⁴C-morphine fell in the presence of DEAE; these results strongly suggest that DEAE could not have been simultanteously inhibiting formation of MES. The reason for the rise in per cent MES is not known. Since TEA is not metabolized during its transtubular transport, the block of its transport by DEAE demonstrates that block of transport by DEAE can occur independently of any possible effect on metabolism of the compound being transported.

The same arguments developed for DEAE can be developed for the action of Lilly 18947 on morphine and TEA transport. However, the supporting experimental evidence is not so strong as that for DEAE, since Lilly 18947 has several additional effects which complicate the interpretation of the results. Namely, Lilly 18947 has a tendency to open the venous bypass valve of the kidney, as indicated by the fall in PAH ATEF. However, Lilly 18947 causes a much greater fall in both ¹⁴C-morphine and ¹⁴C-TEA ATEF values than in the PAH ATEF, so that Lilly 18947 has blocking effects on the organic cation transport system beyond that explicable by opening of the venous bypass valve. This interpretation is consistent with the previous results on SKF 525A and MPDC.¹ A further complication is that Lilly 18947 does appreciably block transport of ¹⁴C-MES administered intravenously. Yet, when the MES is formed from morphine within the renal tubular cell, Lilly 18947 does not appear to block ¹⁴C-MES egress from the cell (Table 1). These two types of MES experiments could be interpreted to mean that Lilly 18947 has some blocking activity against an organic anion transport system known to be located on the peritubular membrane.^{3,4} SKF 525A and MPDC were reported previously not to block transport of administered or intracellularly formed MES.1 By analogy with the latter compounds, the evidence, although weak, is compatible with Lilly 18947 acting to block the transport of organic cations with no inhibition of metabolism of morphine.

As stated in the Introduction, diethylaminoethanol is the portion of the molecule of SKF 525A, Lilly 18947, CFT 1201 and Sch 5712 which confers basic properties to these compounds. Knowing that SKF 525A, diethylaminoethanol itself, and possibly Lilly 18947 block organic cation transport, it would be predicted that CFT 1201 and Sch 5712 should also do the same. Furthermore, other classes of drugs which also contain the diethylaminoethanol moiety, such as procaine, trasentine, and benactyzine, might also block cation transport.

A very interesting study was reported by Foster et al.¹¹ on inhibition of hepatic microsomal enzymes by N-substituted ethanolamines. The ethanolamines including diethylaminoethanol were effective microsomal enzyme inhibitors when administered in vivo. Since broad specificity exists for substrates and inhibitors of microsomal mixed-function oxidases and broad specificity occurs in substrate transport and inhibitor action in the kidney, it is conceivable that the basic mechanism for interaction of certain substrates and inhibitors on the microsomal and renal transport systems might be very similar. The effect of SKF 525A on membranes was discussed in a previous publication.¹

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